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chemical equation (8.1) a way to describe a chemical reaction using chemical formulas and other symbols. precipitate (8.1) a solid that forms from a solution during a chemical reaction. coefficient (8.1)

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precipitate.

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Modern Chemistry Chapter 8. A number in front of a chemical formula in an equation that indicates how many molecules or atoms of each reactant and product are involved in a reaction. When the positive ion of one compound replaces the positive ion of another to form two new compounds.

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Modern Chemistry Chapter 8. STUDY. PLAY. activity series. a list of elements organized according to the ease with which the elements undergo certain chemical reactions. chemical equation. represents, with symbols and formulas, the identities and relative amounts of the reactants and products in a chemical reaction.



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Reactions SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. c synthesis (a)  $AX + BY \rightarrow AY + BX$  d decomposition (b)  $A + BX \rightarrow AX + B$  b single-displacement (c)  $A + B \rightarrow AX$  a double-displacement (d)  $AX \rightarrow A + X$

### *8 Chemical Equations and Reactions*

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Fundamentals of Chemistry, Fourth Edition covers the fundamentals of chemistry. The book describes the formation of ionic and covalent bonds; the Lewis theory of bonding; resonance; and the shape of molecules. The book then discusses the theory and some applications of the four kinds of spectroscopy: ultraviolet, infrared, nuclear (proton) magnetic resonance, and mass. Topics that combine environmental significance with

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descriptive chemistry, including atmospheric pollution from automobile exhaust; the metallurgy of iron and aluminum; corrosion; reactions involving ozone in the upper atmosphere; and the methods of controlling the pollution of air and water, are also considered. Chemists and students taking courses related to chemistry and environmental chemistry will find the book invaluable.

In this fascinating history, Cathy Cobb and Harold Goldwhite celebrate not only chemistry's theories and breakthroughs but

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also the provocative times and personalities that shaped this amazing science and brought it to life. Throughout the book, the reader will meet the hedonists and swindlers, monks and heretics, and men and women laboring in garages and over kitchen sinks who expanded our understanding of the elements and discovered such new substances as plastic, rubber, and aspirin. Creations of Fire expands our vision of the meaning of chemistry and reveals the oddballs and academics who have helped shape our world.

Antoine Lavoisier is considered to be the

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father of modern chemistry. Using experiments and careful measurements, he created a system to help chemists understand how matter behaves. He discovered and named oxygen and hydrogen, and helped set up a system to classify these and other elements. Perhaps his most famous discovery is the role oxygen plays in combustion.

Modern Inorganic Synthetic Chemistry, Second Edition captures, in five distinct sections, the latest advancements in inorganic

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synthetic chemistry, providing materials chemists, chemical engineers, and materials scientists with a valuable reference source to help them advance their research efforts and achieve breakthroughs. Section one includes six chapters centering on synthetic chemistry under specific conditions, such as high-temperature, low-temperature and cryogenic, hydrothermal and solvothermal, high-pressure, photochemical and fusion conditions. Section two focuses on the synthesis and related chemistry problems of highly distinct categories of inorganic compounds, including superheavy elements,

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coordination compounds and coordination polymers, cluster compounds, organometallic compounds, inorganic polymers, and nonstoichiometric compounds. Section three elaborates on the synthetic chemistry of five important classes of inorganic functional materials, namely, ordered porous materials, carbon materials, advanced ceramic materials, host-guest materials, and hierarchically structured materials. Section four consists of four chapters where the synthesis of functional inorganic aggregates is discussed, giving special attention to the growth of single crystals, assembly of nanomaterials,

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and preparation of amorphous materials and membranes. The new edition's biggest highlight is Section five where the frontier in inorganic synthetic chemistry is reviewed by focusing on biomimetic synthesis and rationally designed synthesis. Focuses on the chemistry of inorganic synthesis, assembly, and organization of wide-ranging inorganic systems Covers all major methodologies of inorganic synthesis Provides state-of-the-art synthetic methods Includes real examples in the organization of complex inorganic functional materials Contains more than 4000 references that are all highly reflective of

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chapters on Quantum Mechanics and Molecular Structure (Chapter 6), Electrochemistry (Chapter 17), and Molecular Spectroscopy and Photochemistry (Chapter 20). In addition, the text utilizes mathematically accurate and artistic atomic and molecular orbital art, and is student friendly without compromising its rigor. End-of-chapter study aids focus on only the most important key objectives, equations and concepts, making it easier for students to locate chapter content, while applications to a wide range of disciplines, such as biology, chemical engineering, biochemistry, and medicine deepen students'

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understanding of the relevance of chemistry beyond the classroom.

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THEIRAPPLICATIONS IN CHEMISTRY; AND DEVELOPMENT OF COMPUTATIONAL MODELSFOR ENZYMES, TRANSPORTERS, CHANNELS, AND RECEPTORS RELEVANT TOADME/TOX. FROM REVIEWS OF THE SERIES "Reviews in Computational Chemistry remains the most valuablereference to methods and techniques in computationalchemistry." -JOURNAL OF MOLECULAR GRAPHICS AND MODELING "One cannot generally do better than to try to find an appropriatearticle in the highly successful Reviews in ComputationalChemistry. The basic philosophy of the editors seems to be to helpthe authors produce chapters that are

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complete, accurate, clear, and accessible to experimentalists (in particular) and other nonspecialists (in general)." -JOURNAL OF THE AMERICAN CHEMICAL SOCIETY

This bestselling text introduces descriptive inorganic chemistry in a less rigorous, less mathematical way. The book uses the periodic table as basis for understanding chemical properties and uncovering relationships between elements in different groups. Rayner-Canham and Overton's text also familiarizes students with the historical background of inorganic chemistry as well as with its

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crucial applications (especially in regard to industrial processes and environmental issues), resulting in a comprehensive appreciation and understanding of the field and the role it will play in their fields of further study

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